

469 North Harrison Street Princeton, NJ 08543-5297

800.833.0544 www.ahperformance.com

Specification Sodium Bicarbonate Grade 1 USP

	United States	Food Chemicals Codex	
	Pharmacopoeia		
Description	Sodium Bicarbonate	A white crystalline powder.	
	contains not less than 99.0	It is stable in dry air, but	
	percent and not more than	slowly decomposes in moist	
	100.5 percent of NaHCO ₃	air. Its solutions, when	
	calculated on a dry basis.	freshly prepared with cold	
		water, without shaking, are	
		alkaline to litmus. The	
		alkalinity increases as the	
		solutions stand, are	
		agitated or are heated.	
Identification	A solution of it responds to	A 1 in 10 solution gives	
	the tests for sodium and	positive tests for sodium	
	bicarbonate.	and for bicarbonate.	
Loss on Drying	Not more than 0.25% of its	Not more than 0.25% by	
	weight.	weight	
Insoluble Substances	Dissolve 1 g in 20 ml of	Passes test	
	water; the resulting		
	solution is complete and		
	clear.		
Normal Carbonate	Meets test.	-	
Chloride	150 ppm as Cl Max.	-	
Limit of Sulfur Compounds	150 ppm as SO ₄ Max.	-	
Arsenic	2 ppm Max.	-	
Heavy Metals	5 ppm Max.	-	
Limit of Ammonia	Meets requirements.	Passes test	
Residual Solvents	Meets requirements.	-	
	No residual solvents.		
Assay – dry basis	Not less than 99.0% nor	Not less than 99% NaHCO ₃	
	more than 100.5% of	after drying	
	NaHCO ₃		
Lead	-	Not more than 2 mg/Kg	



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Granulation (Powdered)

		Ro-Tap Cumulative % Retained	
Sieve Size (USS)	Microns	Minimum	Maximum
100	149	0	2
200	74	20	45
325	44	60	100

General Properties (Not Specifications)

Empirical Formula	NaHCO ₃
CAS Number	144-55-8
Other Names	Bicarbonate of Soda
	Sodium Acid Carbonate
	Baking Soda
Chemical Abstract Name	Carbonic acid monosodium salt
Appearance	White crystalline powder
Taste	Slightly alkaline
Molecular Weight	84.01
Thermal Decomposition	Decomposes without melting into Na ₂ CO3,
	H ₂ O and CO ₂ . See Figure 1.
Crystal Density	137.3 lb /ft ³ , 2.2 g / cc
BTU / lb at 72°F	0.249
Solubility in water	See Figure 2.
Solubility in Alcohol	Insoluble
Specific Gravity of Aqueous Solutions	See Figure 3.
Alkali Equivalent	1 lb NaHCO ₃ = 0.369 lb Na ₂ O
Acid Equivalent	1 lb NaHCO ₃ = 0.435 lb HCl
Carbon Dioxide Equivalent	1 lb NaHCO ₃ = 0.524 lb CO ₂
pH 1% aqueous soln at 77°F	Approximately 8.3. See Figure 4.





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